

## 122 Chemical Calculations Worksheet Answer Key

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Molecular weights can be found in the NIOSH Pocket Guide to Chemical Hazards, chemical supplier lists, the NIST Chemistry WebBook or other online databases.. The numeric value of 24.45 in both formulae is the molar volume of air in litres at normal temperature and pressure (NTP), which is considered to be 25°C and 1 atmosphere (101.325 kPa or 760 mm Hg or 760 torr).

Molarity Worksheet # 1

Example 2. Computing Formula Mass for an Ionic Compound Aluminum sulfate, Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>, is an ionic compound that is used in the manufacture of paper and in various water purification processes.What is the formula mass (amu) of this compound? Solution The formula for this compound indicates it contains Al<sup>3+</sup> and SO<sub>4</sub><sup>2-</sup> ions combined in a 2:3 ratio. For purposes of computing a formula mass ...

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3: 122 Average: 130 The student determines that the number of cells in the colony she picked was 130 \* 10<sup>3</sup> \* 10 = 1.3 \* 10<sup>6</sup> where 10<sup>3</sup> is the total dilution factor and 10 is the plating factor.

CHEMISTRY COMPUTING FORMULA MASS WORKSHEET

Chemistry 11 Calculations Practice Test # 1 . 13. Chemistry 11 Calculations Practice Test # 2 . Molarity Worksheet # 1 . 1. 15.8 g of KCl is dissolved in 225 mL of water. Calculate the molarity. 15.8 g x 1 mole

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ChemTeam: Calculations involving molality, molarity ...

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21.1 Nuclear Structure and Stability – Chemistry

(\*)Answer the questions on this Ocean Basin Worksheet. Use your notes and the readings from this unit to help you. Record 10 points for completing it. Take off 1 point for each incomplete answer. Lesson 26(\*) (\*)Answer the questions on this Ocean Floor Worksheet. Use your notes and the readings from this unit to help you.

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3.1 Formula Mass and the Mole Concept – Chemistry

Assume, unless otherwise told, that in all problems water is the solvent. Example #1: Given a density of 1.836 g/mL and a mass percent of H<sub>2</sub>SO<sub>4</sub> of 96.00%, find the molarity, molality, and mole fraction. The molar mass of water is 18.015 g/mol and the molar mass of sulfuric acid is 98.078 g/mol.

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Complete the printed worksheet. Answer all questions and fill in all charts. Record up to 50 points for completing the lab. Lesson 154. Complete page 9 on calculating energy change. Complete page 10 on solving thermochemistry problems. Lesson 155. Go through page 11 on potential energy. Go through page 12 on heating curves. Lesson 156

122 Chemical Calculations Worksheet Answer

X = 90 g O<sub>2</sub> 1 mole O<sub>2</sub> 2 mole KClO<sub>3</sub> 122.5 g KClO<sub>3</sub> = 229.7 g KClO<sub>3</sub> 32 g O<sub>2</sub> 3 moles O<sub>2</sub> 1 mole KClO<sub>3</sub> I II III IV CHEMISTRY STOICHIOMETRY Mass (g,kg, etc.) Volume (L, mL, etc.) or # of Items (atoms,molec.) of Given reactant or product Moles of Given Mole Ratio from the Balanced Chemical Equation

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Nuclear chemistry is the study of reactions that involve changes in nuclear structure. The chapter on atoms, molecules, and ions introduced the basic idea of nuclear structure, that the nucleus of an atom is composed of protons and, with the exception of  ${}^1_1\text{H}$ , neutrons.

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