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10-2 Review and Reinforcements - Front Page

17. The volume of gas particles themselves is assumed to be _____.

18. Gas molecules are said to be in _____. 19. The collisions between gas particles are _____.

20. The temperature of a gas is a measure of the average _____ of the gas ... 12-1 Review and Reinforcement ...

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2 Answer.Key 2-1 Explore, page 5 ...
2-1 Review and Reinforcement, pages 8-9 . 1. energy 2. radiant 3. chemical potential 4. joule 5. created, destroyed ... 17. distillation 18. A magnet could be used to separate them because iron is attracted to a magnet but

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10—2 Review and Reinforcement
(continued) 9. $0.0621 \text{ mol SS} = (Q \$$
 gK_2S 10. $65.8gMg = l. \text{ atoms Mg}$ Solve

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each of the following problems as directed. Show all your work. 11. Calculate the number Of atoms in 0.40 mol of sulfur- 12. Calculate the number of atoms in 2.30 mol of silver. G.oux 13. You have a tank of NOZ gas.

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II—I Review and Reinforcement (continued) Solve each of the following problems as directed. Show all your work. 13. How many moles Of magnesium are required to react with 2.0 mol of hydrochloric acid? The ... 17. Zinc reacts with H_2SO_4 to yield $ZnSO_4$ and hydrogen gas. How manv moles of hydrogen will

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Chemical Families Section 6-3.... KEY
CONCEPTS. - ~ Elements within the

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same family of the periodic table have similar properties because they have the same number of valence electrons. M Vocabulary Skills: Applying Definitions On the periodic table below, label each of the following: acid series alkaline earth ...

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11—1 Review and Reinforcement

Stoichiometry Complete the

following sentences by 'filling in the appropriate word or phrase from the

list below. Each word or phrase

maybe used once, more than once, or

not at all reactants molar ratio

quantitative actual particles.

condrption of matter coefficients

subscripts mass

REVIEW and REINFORCEMENT Section 6-3

Reinforcment & Vocabulary Review

Worksheets To jump to a location in

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left. ... 17. Complete this worksheet

after you finish reading Chapter 2,

Section 1. 1. a. Name three cities

shown on the map that have about

the same ... REINFORCEMENT &

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'16—1A Review and Reinforcement, pages 9-10 6. Answers Will Vary.

When the rates Of forward and reverse reactions are equal, equilibrium is established. For the example of a liquid placed in. a container, equilibrium is established when the Equid vaporizes at the same rate that vapor condenses. 7, NO, in reversible reactions at equilibrium,

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