# 6 2 Covalent Bonding Answers

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6 2 Covalent Bonding Answers
Generally, the strongest types of chemical bonds are the ionic and covalent bonds. Chemical bonds are said to be covalent bond if the bond formed is a result of sharing of electrons between nuclei. Ionics bonds are formed from the mutual electrostatic attraction between oppositely charged ions in a chemical compound.

Ionic Bond (Electrovalent Bond) - Definition, Properties ...
Covalent Character in Ionic Compounds Fajan's Rule. Although atomic bond in a compound like M + X-is considered to be 100% ionic, actually it also has some covalent character. An explanation for the partial covalent character of an ionic bond has been given by Fajan.

Which Bond Is Stronger Ionic Or Covalent - Criteria and ...
Covalent Bond: The ionic bond is the attraction between positive and negative ions in a crystal and compounds held together by ionic bonds are called ionic compounds. The covalent bond is a bond formed when two atoms share one or more electron pairs. Each atom contributes an equal number of electrons towards the bond formation.

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