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Analyzing Chemical Formulas Aim • to perform calculations based on the information in a chemical formula Notes

Determining Formula Mass n add up the product of the subscript and the mass from the periodic table for each element shown in the formula qExamples p silver nitrate

AgNO_3 Ag = $1 \times 108 = 108$ N = $1 \times 14 = 14$ O = $3 \times 16 = 48$
170

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Combined Gas Law Problems 1) A sample of sulfur dioxide
occupies a volume of 652 mL at 40. ° C and 720 mm Hg.

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What volume will the sulfur dioxide occupy at STP?

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proportional $[V \propto 1/P]$. Charles law says at constant
pressure, the volume and temperature of a sample of gas
are directly proportional $[V \propto T]$. Gay-Lussac ' s law says at

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