

## 8th CI Maths State Syllabus

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### **State Functions in Thermodynamics - Definition, Examples ...**

Thus, when an aqueous solution of NaCl containing Na<sup>+</sup>, Cl<sup>-</sup>, H<sup>+</sup> and OH<sup>-</sup> ions is electrolysed, H<sup>+</sup> ion is preferentially deposited at cathode (reduction) instead of Na<sup>+</sup> being reduced, this is

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because reduction potential of hydrogen (0.00 volt) is higher than the reduction potential of sodium (-2.71 volt).

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What is State Function in Thermodynamics? A property whose value doesn't depend on the path taken to reach that specific value is known as state functions or point functions. In contrast, those functions which do depend on the path from two points are known as path functions. State functions are the values which depend on the state of the substance like temperature, pressure or the amount ...

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