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When required to prepare 250 cm³ of 0.5 mol/dm³ solution of a crystal of Mr 50g. First realize that we need only 250 cm³, not 1 dm³ Now in 250 cm³ there will

be: $50/4 = 12.5$ g of the solid. So, first add 12.5 g of solid to 50 cm³ of water in a BEAKER (not the volumetric flask yet). Also note that the volume of water is less than 250 cm³.

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for you to be able to work p5 chemistry, it is necessary to be good in practical(p3) because p5 test your practical knowledge. you may have noticed that some question that appears in p3 are modified into a p5 question. so before working a p5 , look at similar question in a p3. most of the time you get the p5 answer in the p3 of similar experiment.

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