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Notes ACIDS, BASES AND SALTS

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Salt is an ionic compound that results from the neutralization reaction of acids and bases. Salts are constituted of positively charged ions, known as cations and negatively charged ions, known as anions, which can either be organic or inorganic in nature.

Acids, Bases and Salts Class 10 Important Questions with ...

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Acids, Bases and Salts SCIENCE AND TECHNOLOGY

Notes MODULE - 2 Matter in our Surroundings

Acid Base INTEXT QUESTION 8.1 1. Put the following substances in acid or base bottle.

(a) Milk of magnesia (b) gastric juice in humans (c) soft drinks (d) lime water (e) vinegar (f) soap 2. What will happen if you add a drop of the following on a cut ...

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Acids, Bases and Salts: Experimental
Definitions ...

Reaction of Acids and Bases with each other •

Acids and Bases react to form salt and water.

Acid + Base ? Salt + H₂O • Neutralisation

Reaction: Reaction of acid with a base is
called as neutralization reaction. Example:

HCl + NaOH ? NaCl + H₂O • Strong Acid +

Weak Base ? Acidic salt + H₂O • Weak Acid +

Strong Base ? Basic salt + H₂O ...

Acids, Bases and Salts-I

CHEMISTRY ACIDS, BASES & SALTS

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Salts having same positive ions (or same negative ions) are said to belong to a family of salts. pH of Salts Salts of strong acid and a strong base are neutral, with a pH value of 7. For Example: NaCl, Na_2SO_4

Class 10 Chapter 2 Science Notes Acids, Bases, and Salts

From this App you can learn : Define the terms acid, base, amphoteric substance and salt. Classify the substances into acids, bases, amphoteric substances and salts. Differentiate the properties of acids and bases. Appreciate the various theories

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developed to describe acid base behavior of substances. Define pH and express the strength of an acid or base on pH scale.

Acids Bases and Salts Class 10 Notes Science Chapter 2 ...

The salts of strong acids and weak bases give acidic solution having pH less than 7.

Example, NH_4Cl , Ammonium Chloride will have pH less than 7. The salts of weak acids and strong bases give basic solution having pH more than 7. Example, Na_2CO_3 , Sodium Carbonate will have pH more than 7. Question 7.

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Notes Chapter - 2 Acids, Bases and Salts

ACIDS: • These are the substances which have sour in taste. • They turns blue litmus solution to red. • They give H ions in aqueous solution Examples of Acids : - HCl - Hydrochloric Acid BASES • These substances are bitter in taste. • They turns red litmus solution to blue.

Acids, Bases, and Salts - Introduction,
Dissociation ...

Acids, bases and salts affect chemistry as

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well as our day to day life. They can be easily identified by their taste, that is acids taste sour and bases taste bitter and salts itself have salty taste. Read about the examples and uses of Acid, Bases and salts at Vedantu.com.

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Acids, Bases and Salts 1. Select the correct word from the list in brackets for each of the following statements : [1mark each] (i) A more volatile acid obtained, when a less volatile acid reacts with sodium nitrate is

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_____ [sulphuric acid/nitric acid]. Ans.

Nitric ...

Notes of Ch 2 Acids, Bases and Salts | Class
10th Science

Concentrated acid – if the concentration of an acid is high, it is called concentrated acid. Bases : Bases are those chemical substances which have a bitter taste.

According to Arrhenius concept, "are substances that release OH^- ions in an aqueous solution".

Revision Notes on Acids, Bases and Salts -

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Askiitians

In a neutralization reaction, acid and base reacts to form salt and water. Heat is evolved in the reaction. Question 6: What is the color of litmus when added to distilled water? Red. Blue. Mauve. White. Correct Option is : 3. Solution : Lichen is a natural indicator, it gives mauve (purple) colour in distilled water.

MCQ Questions on Acids Bases And Salts Class 10 - LearnFatafat

Acids, Bases and Salts. All substances are acidic, neutral or basic (alkaline). How

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acidic or basic a substance is shown by its pH. There are several other ways by which we could find out whether a substance is acidic, neutral or basic. pH Scale: This is a scale that runs from 0 to 14.

Acids, Bases and Salts - GCSE Chemistry
Revision ...

A) Change of colour in an acid and a base depends on the type of the indicator. B) Both acids and bases change colour of all indicators. C) None of the above D) If an indicator gives a colour change with an acid, it does not give a change with a base. E) If

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an indicator changes colour with a base, it does not change colour with an acid.

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Acids, Bases and Salts

A salt of a strong acid and a strong base will be neutral in nature. $\text{pH} = 7$ (approx.).

A salt of a weak acid and a strong base will be basic in nature. $\text{pH} > 7$. A salt of a strong acid and a weak base will be acidic in nature. $\text{pH} < 7$. The pH of a salt of a weak acid and a weak base is determined by conducting a pH test. Preparation of Sodium

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Acids Bases and Salts Chapter Wise Important Questions ...

Earlier, acids, bases, and salts were characterized by the experimental testing of their aqueous solutions. An acid is defined as a substance whose water solution tastes sour, turns blue litmus red and neutralizes bases. A substance is called base if its aqueous solution tastes bitter, ...

Acids Bases and Salts notes for CBSE Class 10 Chemistry

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Acids Bases and Salts | Properties of Acids, Bases and Salts

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ACIDS, BASES, SALTS - Testlabz

The substances which react with acids to form salt and water as the only products are called bases. All bases which are soluble in

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water are called alkalis. 3. A salt formed by the complete replacement of the hydrogen ions (H^+) of an acid with metal ions is called a neutral salt. When a salt is formed by the partial replacement of H^+ ions of an ...

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