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Answers To Explore Learning Calorimetry
DESCRIPTION. Investigate how calorimetry can be used to find

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relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs.

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Calorimetry: The science of measuring heat flow that occurs during a physical or chemical change. Heat: Form of energy that moves from one object to another because of a difference in temperature. When a hot object is placed in contact with a cold object heat flows from the hot object to the cold object until they reach?

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Student Exploration: Calorimetry

Lab Vocabulary: calorie, calorimeter, joule, specific heat capacity
Prior Knowledge

Questions (Do these BEFORE using the Gizmo.) 1. The Latin word calor means “heat,” and meter comes from the Greek word meaning “to measure.” What do you think a calorimeter does?

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lost by the other substance. Suppose you place 125 g of aluminum in a calorimeter with 1,000 g of water. The water changes temperature by $2\text{ }^{\circ}\text{C}$ and the aluminum changes temperature by $-74.95\text{ }^{\circ}\text{C}$. A.

Student Exploration: Calorimetry

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Example Question #1 :

Calorimetry, Specific Heat, And Calculations. 1. The density of water allows us to say that 300 milliliters of water is the same thing as 300 grams of water. 2. Since the heat from the iron is being transferred to the water, we can say that the heat transfer is equal between both compounds.

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