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b. The planet Earth is not a closed system. Ultimately, what is the source of all energy for life on Earth? 23. Consider a process where the increase in entropy of the system is exactly equal to the decrease in entropy of the surroundings. Would that process be spontaneous? Justify your reasoning.  
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Paul Andersen attempts to explain Gibbs Free Energy. He begins by using three spontaneous reactions to explain how a change in enthalpy, entropy and temperature can affect the free energy of a system. He then applies this concept to cellular respiration and photosynthesis.

HOMEWORK - AP BIOLOGY

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Free Energy 1 Free Energy Where does the energy come from that keeps all living things alive? Why? ... 2™ Activities for AP\* Biology POGIL 2. Refer to Model 1. For each process below indicate if the change would be endothermic or exothermic. a. A plant making glucose from carbon dioxide and water. b.

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