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while metallic bonds have lower bonding energies than covalent or ionic materials. 2-21 What type of bonding does KCl have? Fully explain your reasoning by referring to the electronic structure and electronic properties of each element. Solution: KCl has ionic bonding. The electronic structure of [K] =  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^1$  = [Ar]  $4s^1$ . The

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