

## Basic Chemistry Mcq With Answers

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Science Quiz Chemistry Quiz | Basic Chemistry Questions – Multiple Choice (MCQ) from 16 to 20. CHEMISTRY QUIZ. Page: 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10. 16. NH<sub>3</sub> is a/an (A) Acid (B) Base (C) Salts (D) Neutral. 17. Salts are (A) Acid (B) Base (C) Alkalies (D) Neutral. 18. Organic compounds are originated from ? (A) Air (B) Sun (C) Rocks (D) Living organisms

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A. Isotopes with the odd atomic number are abundant. B. Isotopes with the odd atomic number and even mass number are abundant. C. Isotopes with an even atomic number and even mass number are abundant. D. Isotopes with an even atomic number and odd mass no are abundant. View Answer. Answer: Option C. Click On Below Page Numbers To Go To Next/Previous Page of Chemistry Basic Concept MCQs.

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Answer: (b) 2. Which one will have maximum numbers of water molecules? (a) 18 molecules of water (b) 1.8 grams of water (c) 18 grams of water (d) 18 moles of water. Answer: (d) 3. The number of atoms present in 0.1 moles of a triatomic gas is (a) 1.806 × 10<sup>23</sup> (b) 1.806 × 10<sup>22</sup> (c) 3.600 × 10<sup>23</sup> (d) 6.026 × 10<sup>22</sup>. Answer: (a) 4.

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Organic Chemistry: Some Basic Principles and Techniques Class 11 MCQs Questions with Answers. Question 1. If two compounds have the same empirical formula but different molecular formula they must have (a) Different percentage composition (b) Different molecular weight (c) Same viscosity (d) Same vapour density. Answer. Answer: (b) Different molecular weight

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Answer & Explanation Answer: C) Bromine Explanation: The chemical element with the atomic number 35, Bromine is a non metal that remains liquid at room temperature. Mercury also remains liquid at room temperature but it is a metal.

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Atomic weight = (Equivalent weight × Valency) = (8.9 × 3) = 26.7. (Valency = (26.89)/ (8.9) 3). Hope the information shared regarding the NCERT MCQ Questions for Class 11 Chemistry Chapter 1 Some Basic Concepts of Chemistry with Answers Pdf free download is true and genuine as far as our knowledge is concerned.

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