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What are the Two Types of Ions and how are they Different Ion: An ion is a positively or negatively charged atom (or group of atoms). An ion is formed by the loss or gain of electrons by an atom, so it contains an unequal number of electrons and protons. Example: Sodium ion Na^+ , magnesium ion [...]

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Definitions with examples : 1. Combination Reaction: Two or more reactant combine to form a single product. $2 \text{Mg (s)} + \text{O}_2 \text{ (g)} \rightarrow 2 \text{MgO}$ Magnesium Oxygen Magnesium oxide (White ash) (basic) turns Red litmus blue. 2. Decomposition Reaction: A single compound decomposes or break down to give two or more simpler substances. $2 \text{FeSO}_4 \text{ (s)}$.

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