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NCERT Solutions Class 12 Chemistry Chapter 5 Surface ...

*Definitions with examples : 1. Combination Reaction: Two or more reactant combine to form a single product. $2 \text{Mg (s)} + \text{O}_2 \text{(g)} \rightarrow 2 \text{MgO}$
Magnesium Oxygen Magnesium oxide (White ash) (basic) turns Red litmus blue. 2. Decomposition Reaction: A single compound decomposes or break down to give two or more simpler substances. $2 \text{FeSO}_4 \text{(s)}$*

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*NCERT Solutions for Class 12 Chemistry Chapter 5 Surface Chemistry is an imperative study material that is required for the perfection of surface chemistry topics. These solutions for class 12 Chemistry Chapter 5 pdf provided here has answers to the questions in the textbooks, NCERT exemplar problems, MCQS, short and long answer questions. *The complete and accurate NCERT Solutions for Class ...*

CBSE 9, Chemistry, CBSE- Atoms and Molecules, Notes

What are the Two Types of Ions and how are they Different Ion: An ion is a positively or negatively charged atom (or group of atoms). An ion is formed by the loss or gain of electrons by an atom, so it contains an unequal number of electrons and protons. Example: Sodium ion Na^+ , magnesium ion [...]

chemical reactions and equations cbse notes for class 10 ...

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