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A mixture of a weak acid and its conjugate base (or a mixture of a weak base and its conjugate acid) is called a buffer solution, or a buffer. Buffer solutions resist a change in pH when small amounts of a strong acid or a strong base are

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added (). A solution of acetic acid and sodium acetate ($\text{CH}_3\text{COOH} + \text{CH}_3\text{COONa}$) is an example of a buffer that consists of a weak acid and its salt.

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Heterogeneous

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Explain Use the properties of seawater to describe the characteristics of mixtures. Answers will vary but might include that seawater is a heterogeneous mixture with dirt

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