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___ 10. In an exothermic reaction, the energy stored in the chemical bonds of the reactants is ____. a. equal to the energy stored in the bonds of the products b. greater than the energy stored in the bonds of the products c. less than the energy stored in the bonds of the products d. less than the heat released ___ 11.

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CHEMISTRY: Chapter 10 Prep-Test

The phrase "like dissolves like" refers to the fact that ____. a) polar solvents dissolve nonpolar solutes and vice versa b) solvents can only dissolve solutes of similar molar mass

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**Chapter 10 test Matching Match each item with the correct statement below. a. representative particle d. percent composition b. mole e. standard temperature and pressure c. Avogadro's number f. empirical formula
___ 1. the number of representative particles of a substance present in 1 mole of that substance**

CBSE 10, Chemistry, CBSE- Chemical Reactions and Equations ...

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