

Chapter 10 Physical Characteristics Of Gases

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attractive forces between them. In Figure 10-2, hydrogen gas diffuses rapidly into other gases at the same temperature because its molecules are lighter and move faster than the molecules of the other gases. PHYSICAL CHARACTERISTICS OF GASES 305 Stopcock closed Stopcock open H 2 N 2 O 2 FIGURE 10-2 When hydrogen gas in a flask is allowed to mix ...

CHAPTER 10 Physical Characteristics of Gases

Chapter 10 - Physical Characteristics of Gases 10-1 The Kinetic-Molecular Theory of Matter I. The Kinetic-Molecular Theory of Gases A. Ideal Gas 1. An imaginary gas that perfectly fits all the assumptions of the kinetic-molecular theory B. Five Assumptions of the Kinetic-Molecular Theory 1.

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While the sequence of physical changes in puberty is predictable, the onset and pace of puberty vary widely. Several physical changes occur during puberty, such as adrenarche and gonadarche, the maturing of the adrenal glands and sex glands, respectively. Also during this time, primary and secondary sexual characteristics develop and mature.

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