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(48) chemical equation. a symbolic way of
describing a chemical reaction. skeleton
equation. a chemical equation that does not
indicate the amounts of substances involved.
catalyst. a substance that speeds ...**

**11.1 Describing Chemical Reactions 11
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**Chapter 11 Packet - Humanlyspeaking,
destroyed.,bothsides ...**

**11.2 Types of Chemical Reactions > 29 Whether
one metal will displace another metal from a
compound depends upon the relative
reactivities of the**

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Section 11.1 - Describing Chemical Reactions In
a chemical reaction, the reactants are written**

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on the left and the products on the right. The arrow that separates them is called yield.

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DESCRIBING CHEMICAL ...

Chapter 11 - CHEMICAL REACTIONS Law of Conservation of Matter: Humanly speaking, matter is neither created nor destroyed...therefore in a chemical reaction, both sides (reactants and products) must contain the same number of atoms of each element.

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Problems Answer Key ...

Chapter 11 Chemical Reactions 265 ... Answer the following in the space provided. 22. Identify the type of each of the following reactions. a. $2\text{C} + 6\text{H} \dots$ SECTION 11.1 DESCRIBING CHEMICAL REACTIONS 1. Write the skeleton equation for the reaction between hydrogen and oxygen that

11.2 Types of Chemical Reactions >

330 Chapter 11 11.2 Types of Chemical Reactions Often charcoal briquettes ... Chemical Reactions 331 Answers to... Figure 11.5 The reaction produces bright, white light. Gifted and Talented Combination reactions are sometimes referred to as synthesis reactions, but synthe-

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Chapter 11

Although the chemical reactions involved in photosynthesis and digestion are different,

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both chemical reactions are necessary to sustain life. In a chemical reaction, one or more reactants change into one or more products. Figure 11.1a shows the ingredients for making leavened bread— flour, salt, yeast, and water.

**Chapter 11: Chemical Reactions
Chapter 11 ... 1.**

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