

Chapter 11 Chemical Reactions

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Chapter 11 – CHEMICAL REACTIONS Law of Conservation of Matter: Humanly speaking, matter is neither created nor destroyed...therefore in a chemical reaction, both sides (reactants and products) must contain the same number of atoms of each element.

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329)

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CHAPTER Date STUDY GUIDE Class Chemical Reactions Section 9.1 Reactions and Equations In your textbook, read about evidence of chemical reactions. For each statement, write yes if evidence of a chemical reaction is present. Write no if there is no evidence of a chemical reaction. 2. 4. 5. 6. 8. A tomato smells rotten.

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Chapter 11 - Chemical Reactions - Preston Treend

Chemical Reactions (Chapter 11) By C B 6th period . 1) Combination Reactions • Is also referred to as a synthesis reaction • It is a chemical change in which two or more substances react to form a new singular substance • The product is a compound in this form of reaction

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Chapter 11. Chemical Equations - University of New Mexico

Chapter 11 Chemical Reactions113 SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329) This section explains how to write equations describing chemical reactions using appropriate symbols. It also describes how to write balanced chemical equations when given the names or formulas of the reactants and products in a chemical reaction.

The 5 Types of Chemical Reactions (Chapter 11)

142 Chapter 11 Chemical Equations products of chemical reactions showed that elements are neither destroyed nor created when substances are.These observations, coupled with the conservation of mass observed for reactions, led John Dalton to surmise that atoms arenot destroyed nor created during chemi-

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a representation of a chemical reaction with formulas of reactants to the left of an arrow and the formulas of the products to the right of an arrow combination (synthesis) reaction chemical change in which two or more substances react to form a single new substance

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A double arrow, \rightleftharpoons , shows that the reaction is reversible and can go in both directions. 5. Before beginning to balance an equation, check each formula to see that it is correct.

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Section 11.1 – Describing Chemical Reactions In a chemical reaction, the reactants are written on the left and the products on the right. The arrow that separates them is called yield. Reactants ? Products

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