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SHORT ANSWER

Answer the following questions in the space provided.

c c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (b) divided by the mass of 1 mol. nRT (c) multiplied by 22.4 L. (d) divided by 22.4 L.

For the expression  $V =$

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(a)

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SECTION 3 SHORT

ANSWER Answer the  
following questions in  
the space provided 1

The molar mass of a  
gas at STP is the  
density of that gas (a)



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multiplied by the mass  
of 1 mol (c) multiplied  
by 224 L (b) divided by  
the mass of 1 mol (d)  
divided by 224 L 2 For  
the expression  $V = nR \frac{P}{T}$

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SHORT ANSWER .  
Answer the following

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questions in the space provided. 1. Consider the following data table: Approximate pressure (kPa)

Altitude above sea level (km)	100	0 (sea level)	50	5.5 (peak of Mt. Kilimanjaro)	25	11 (Get cruising altitude)	< 0.1	22 (ozone layer)	~.
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Answer the following 9 questions in the space provided. 1. c The molar mass of a gas at STP is the density of

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that gas (a) multiplied  
by the mass of 1 mol.  
(c) multiplied by 22.4 .  
L. (b) divided by the  
mass of 1 mol. (d)  
divided by 22.4 . L. 2. c  
For the expression  $V =$   
 $n \cdot T,$

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#### SHORT ANSWER

Answer the following questions in the space provided.

c c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (b) divided by the mass of 1 mol.  $nRT$  (c) multiplied by 22.4 L. (d) divided by 22.4 L.

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Answer the following 9 questions in the space provided. 1. c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L. (b) divided by the mass of 1 mol. (d) divided by 22.4 L. 2. c For the ...

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(Section 10.2) Review  
Skills The presentation  
of information in this

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chapter assumes that you can already perform the tasks listed below. You can test your readiness to proceed by answering the Review Questions at the end of the chapter.

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Gases Mixed Answers  
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## SECTION 2 SHORT

ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances: a. temperature increases, volume stays the same b. volume increases,

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temperature stays the same

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## SECTION 3 SHORT

ANSWER Answer the following questions in the space provided 1 The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol (c) multiplied by 224 L (b) divided

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SHORT ANSWER

Answer the following questions in the space provided.

c c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (b) divided by the mass of 1 mol.  $nRT$  (c)

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multiplied by 22.4 L.

(d) divided by 22.4 L.

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REVIEW Molecular

Composition of Gases

MIXED REVIEW

SHORT ANSWER

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Answer the following questions in the space provided. 1. The average speed of a gas molecule is most directly related to the .  
(a) polarity of the molecule (b) pressure of the gas (c) temperature of the gas (d) number of moles in ...

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11 REVIEW Gases  
SECTION 3 SHORT  
ANSWER Answer the  
following questions in  
the space provided 1  
The molar mass of a  
gas at STP is the  
density of that gas (a)  
multiplied by the mass

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of 1 mol (c) multiplied  
by 224 L (b) divided by  
the mass of 1 ...

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SHORT ANSWER

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Answer the following questions in the space provided.

c c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (b) divided by the mass of 1 mol.  $nRT$  (c) multiplied by 22.4 L. (d) divided by 22.4 L.

For the expression  $V =$  (a) increasing  $P$  (b) decreasing  $T$

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