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Chapter 12 Multiple Choice Identify the letter of the choice that best completes the statement or answers the question. ____ 1. The first step in most stoichiometry problems is to _____. a. add the coefficients of the reagents c. convert given quantities to volumes b. convert given quantities to moles d. convert given quantities to masses ____ 2.

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Chapter 12 Stoichiometry Section 12.1 The Arithmetic of Equations Using Balanced Chemical Equations Chemists use balanced chemical equations as a basis to calculate how much reactant is needed or product is formed in a reaction. Stoichiometry = calculation of quantities in chemical reactions is a subject of chemistry.

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In Example 12.2.1 and Example 12.2.2, the identity of the limiting reactant has been apparent: [Au(CN) 2]?, LaCl 3, ethanol, and para-nitrophenol. When the limiting reactant is not apparent, we can determine which reactant is limiting by comparing the molar amounts of the reactants with their coefficients in the balanced chemical equation, just as we did in Section 11.4 .

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