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***Chapter 11 study guide True/False Indicate whether the statement is true or false. _____
1. The actual yield is always lower than the theoretical yield. Matching Match each item with the correct statement below. ? a. stoichiometry ? ? b. mole ratio ? ? c. stoichiometric equivalent ? _____ 2.***

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STUDY GUIDE Date Class Stoichiometry Section 11.1 What is stoichiometry? ... 12. 13. 14. 72 Methanol Oxygen gas Carbon dioxide Water What are the reactants? C) What are the products? 10B ... Chemistry: Matter and Change Chapter II ...

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In your textbook, read about why reactions stop and how to determine the limiting reactant. Study the diagram showing a chemical reaction and the chemical equation that represents the reaction.

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CHAPTER Section 11.1 continued In your textbook, read about mole ratios. Answer the questions about the following chemical reaction. sodium + iron(III) oxide → sodium oxide + iron
 $6\text{Na}(s) + \text{Fe}_2\text{O}_3(s) \rightarrow 3\text{Na}_2\text{O}(s) + 2\text{Fe}(s)$
15. What is a mole ratio? ... Study Guide Chemistry: Matter and Change Chapter 11 ...

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TEACHER GUIDE AND ANSWERS Study Guide - Chapter 11 – Stoichiometry Section 11.1 What is stoichiometry? 1. true 2. true 3. false 4. true 5. true 6. 2, 2, 64.10 7. 3, 3, 96.00 8. 2, 2, 88.02 9. 4, 4, 72.08 10. methanol and oxygen gas 11. carbon dioxide and water 12. 160.10 g 13. 160.10 g 14. They are equal. 15. A mole ratio is a ratio between ...

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