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Chapter 12 Multiple Choice Identify the letter of the choice that best completes the statement or answers the question. \_\_\_\_ 1. The first step in most stoichiometry problems is to \_\_\_\_, a. add the coefficients of the reagents c. convert given quantities to volumes b. convert given quantities to moles d. convert given quantities to masses \_\_\_\_ 2.

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e. 12 g 7. How many grams of nitric acid, HNO<sub>3</sub>, can be prepared from the reaction of 138 g of NO<sub>2</sub> with 54.0 g H<sub>2</sub>O according to the equation below? 3NO<sub>2</sub> + H<sub>2</sub>O → 2HNO<sub>3</sub> ... Practice Test Ch3 Stoichiometry (page 3 of 3) 1. d It might be easiest to balance the equation with mostly whole numbers: 2 NH<sub>3</sub> + 3 O<sub>2</sub> → 2 NO<sub>2</sub> + 3 H<sub>2</sub>O. The ...

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