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Chapter 13 Answer Key

Equilibrium cannot be established between the liquid and the gas phase if the top is removed from the bottle because the system is not closed; one of the components of the equilibrium, the Br 2 vapor, would escape from the bottle until all liquid disappeared. Thus, more liquid would evaporate than can condense back from the gas phase to the liquid phase.

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chapter 13 answer key. 1. How can energy reserves make a population experience delayed density dependence? a. They enable the population to survive above carrying capacity. b. They protect organisms from devastating events like hurricanes or floods. c.

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