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covered by this chemistry handout. It serves as a review of a specific textbook section, but will also serve as a nice review for any general chemistry class.

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A vodcast I made for watching later in the chapter (after we have talked about strong and weak acids and bases in class as well as salts). It goes over all the different categories of pH problems in this chapter (strong acid, weak base, salt of weak acid/strong base, etc) and how to solve them.

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acid and base compounds and their reactions is presented with this handout. Pupils name compounds and write chemical formulas.

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review of acids and bases. It begins by addressing conjugate pairs and acid base reactions.

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Figure 14.8. Relative Strength of Acids

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& Bases 25 Conjugate Acid-Base Pair
Trends 1. A stronger acid loses its proton more readily than a weaker acid and a stronger base gains a proton more readily than a weaker base. 2. The stronger the acid, the weaker its

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Acids and Bases: Chapter 14 & 15 .

HW: •Read Ch 14: •Fill in as much of the acid base table as you can, as you read . Acid base conductivity and reactivity Conductivity, Reac'tivity,, Hydrochloric*acid* high high

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Acidic*acid* ** low* medium

Dis4lled*water* none* none*

Ammoniumhydroxide med* none*

Sodiumhydroxide* high none* *

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