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- 15.1 Lesson Check -

Page 493: 1 Answer The

hydrogen bonding in  
water cause the high  
surface tension and low  
vapor pressure of water.

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Worksheet Answers ...

1. Chapter 15 “ Water  
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Pre-AP Chemistry

Charles Page High

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School Stephen L.  
Cotton 2. Section 15.1

Water and it ' s

Properties

OBJECTIVES:

– Explain the high surface tension and low vapor pressure of water in terms of the structure of the water molecule and hydrogen bonding.

3.

Water And Aqueous

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ISBN-10: 0132525763,

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SECTION 15.1

WATER AND ITS

PROPERTIES (pages  
445 – 449) This section  
describes the properties  
of water in the liquid  
and solid states and  
explains how hydrogen  
bonding affects the  
surface tension and  
vapor pressure

Chapter 15 - Water and

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Chapter 15 (Water and  
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Study Guide The  
bonds between the  
hydrogen and oxygen  
atoms in a water  
molecule are polar  
covalent bonds. The  
covalent bonds are polar  
because the oxygen  
atom has a greater  
electronegativity than  
the hydrogen atoms.

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The bonds between adjacent water molecules are called hydrogen bonds.

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WATER AND

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chapter ... Crystals of  
copper sulfate

pentahydrate always

contain five molecules of  
water for each copper

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and sulfate ion pair.

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use these activities to  
yourself study the  
vocabulary and major  
concepts presented in  
this chapter

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Worksheet Answers —

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## Water Aqueous

Assessment ...

The high surface tension of water is due to the: a.

small size of water

molecules. b. low mass of

water molecules. c.

hydrogen bonding

between water

molecules. d. covalent

bonds in water

molecules. \_\_\_\_\_ 12.

Salts and other

compounds that remove

moisture from air are

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Water, Aqueous

said to be: a.  
efflorescent. c. colloidal.  
b. surfactant. d.  
hygroscopic. 10 9 ...

SECTION 15.1  
WATER AND ITS  
PROPERTIES (pages  
445 – 449)

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and aqueous systems.

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AQUEOUS ...

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Behavior of Gases

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Bases and Salts Chapter

20 - Oxidation-

Reduction Reactions

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Aqueous ...

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#### Water Aqueous

Assessment - Page 511:

64 Answer Anions are

attracted to the

hydrogen atoms of the water molecule, because hydrogen atoms are slightly positive.

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"Water and Aqueous Systems"

Chapter 15 - Water and Aqueous Systems - 15.3

Heterogeneous Aqueous



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Systems - 15.3 Lesson

Check - Page 507: 21

Answer Colloids cannot be separated through filtering because their particles are smaller than those of suspensions and they do not settle out over time.

Chapter 15 Water

Aqueous Systems

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