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Page 387 WATER AND AQUEOUS

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Chapter 13 - States of
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and Aqueous Systems Chapter
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-Thermochemistry Chapter 18
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Chapter 15 - Water and Aqueous Systems - 15 Assessment ...

The high surface tension of
water is due to the: a.
small size of water
molecules. b. low mass of
water molecules. c. hydrogen
bonding between water
molecules. d. covalent bonds
in water molecules. _____

12. Salts and other
compounds that remove
moisture from air are said
to be: a. efflorescent. c.
colloidal. b. surfactant. d.
hygroscopic. 10 9 ...

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SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449)

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1. Chapter 15 "Water and Aqueous Systems" Pre-AP Chemistry Charles Page High School Stephen L. Cotton 2. Section 15.1 Water and it's Properties OBJECTIVES:
-Explain the high surface tension and low vapor pressure of water in terms of the structure of the water molecule and hydrogen bonding. 3.

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Chapter 15 - Water and Aqueous Systems - 15.3 Heterogeneous ...

Chapter 15 - Water and
Aqueous Systems - 15.1 Water
and Its Properties - 15.1
Lesson Check - Page 493: 1
Answer The hydrogen bonding
in water cause the high
surface tension and low
vapor pressure of water.

Chapter 15 water and aqueous systems - SlideShare

WATER AND AQUEOUS SYSTEMS
chapter ... Crystals of
copper sulfate pentahydrate
always contain five
molecules of water for each
copper and sulfate ion pair.

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Quia - Chapter 15 "Water and Aqueous Systems"

Chapter 15 - Water and

Aqueous Systems - 15.3

Heterogeneous Aqueous Systems

- 15.3 Lesson Check - Page

507: 21 Answer Colloids

cannot be separated through

filtering because their

particles are smaller than

those of suspensions and

they do not settle out over

time.

Chapter 15 - Water and Aqueous Systems - Standardized Test ...

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Chapter 15 - Water and Aqueous Systems - 15.1 Water

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Chapter 15 (Water and Aqueous Systems) Test Study Guide ? The bonds between the hydrogen and oxygen atoms in a water molecule are polar covalent bonds. ? The covalent bonds are polar because the oxygen atom has a greater electronegativity than the hydrogen atoms. ? The bonds between adjacent water molecules are called hydrogen bonds.

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Answers

Prentice Hall Chemistry

**Chapter 15: Water and
Aqueous ...**

Chapter 15 - Water and

Aqueous Systems - 15

Assessment - Page 511: 64

Answer Anions are attracted to the hydrogen atoms of the water molecule, because hydrogen atoms are slightly positive.

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aqueous systems Flashcards
and ...**

Chapter 15 Water and Aqueous
Systems 159 SECTION 15.1

WATER AND ITS PROPERTIES

(pages 445-449) This section describes the properties of water in the liquid and solid states and explains

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how hydrogen bonding affects
the surface tension and
vapor pressure

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concepts presented in this
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