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1 . The instantaneous rate is the rate of a reaction at any particular point in time, a period of time that is so short that the concentrations of reacta

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Chapter 17 a ntroduction o B synthetIC p

In basic solution, [OH ?] > 1 × × 10 ?? M > [H +].Hydrogen ion cannot appear as a reactant because its concentration is essentially zero. If it were produced, it would instantly react with the excess hydroxide ion to produce water.

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Question 6 The following graph shows the stability of a molecule as its structure is varied during conformational analysis. If energy minimisation was carried out on a starting structure appearing as shown on the graph, at which point would energy minimisation stop?

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