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SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK

(pages 505–510)

CHAPTER 17,
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GUIDED PRACTICE
PROBLEM 12 (page
513) 12. When
50.0 mL of water
containing 0.50

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mol HCl at
22.5°C is mixed
with 50.0 mL of
water containing
0.50 mol NaOH at
22.5°C in a
calorimeter, the
temperature of
the solution
increased to
26.0°C.

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CHAPTER 17, Therm
ochemistry
(continued) 6.

In
thermochemical
calculations, is
the direction of
heat flow given

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from the ...
thermochemical
equation in the
first paragraph
on page 517 as a
guide. SECTION
17.3 HEAT IN
CHANGES OF STATE
(pages 520–526)

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CHAPTER 17,

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(continued) 2.

What is

calorimetry?

Calorimetry is
the accurate and
precise

measurement of
heat change for
chemical and
physical
processes. 3.

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Use Figure
17.5 on page 511.

Circle the letter next to each sentence that is true about calorimeters. G

The calorimeter container is insulated to minimize loss of heat to or

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Matching Match

each item with
the correct
statement below.

- a. calorimeter
- d. enthalpy
- b. calorie
- e. specific heat
- c. joule
- f. heat capacity
- _____ 1. quantity of heat needed to raise

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the temperature
of 1 g of water

Answers _____ 2.

SI unit of

energy _____ 3.

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$Q = m C \Delta T$
 $-q C = 4.184 \text{ J}$
moles g $^{\circ}\text{C}$ 1.

When 150-g
sample of KCl
dissolves in
65.0 g of water
in a
calorimeter, the
temperature

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Answers
Name _____ Date

_____ Period

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- 17.2

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Equations 1.

Make the

following

conversions: a.

444 cal to

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joules = 1.86×10^3 J b. 1.8 kJ

to joules = 1.8×10^3 J c. 0.45

kJ to calories =
 1.1×10^2 cal 2.

Classify each of
these processes
as endothermic
or exothermic:

a. condensing
steam ...

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Terms in this

set (55) Energy.

The capacity for
doing work or
supplying heat.

Thermochemistry.

The study of
energy changes
that occur
during chemical
reactions and
changes in
state.

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Thermochemistry
is the
scientific study
of the heat
energy that is
involved in or

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produced by
chemical

reactions and/or
physical
transformations.

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tell us about
how this topic
is studied and
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thermochemistry.

The study of energy changes that occur during chemical reactions and changes in state. chemical potential energy. The energy stored in

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15.1 – 15.2

Answers

Heat,
Calorimetry, and
Enthalpy. Make
the following
conversions: 444
cal to joules .
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