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Acids produce what type of ions? H^+ Acids taste ____
Acids turn ____ litmus paper _____. 1) sour 2) blue, red pH
of an acid is ____ than 7. Less What are two properties of
strong acids? Good conductors and strong electrolytes
Give an example of at least one strong acid. Any of the
following: [...]

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Chapter 19 "Acids, Bases, and Salts ... Figure 19.22, page 615. Neutralization. The solution of known concentration is called the standard solution. added by using a buret. Continue adding until the indicator . changes color.

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Section 19.1 – Acid-Base Theories Acids have a sour taste, change the color of an indicator, can be strong or weak electrolytes in aqueous solution, and react with metals.

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A B; hydrolyzing salts: compounds derived from the reaction of a strong base with a weak acid or a strong acid with a weak base or a solution of a weak base and one of its salts

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594 Chapter 19 Acids and Bases CHAPTER 19 What You'll Learn You will compare acids and bases and understand why their strengths vary. You will define pH and pOH and calculate the pH and pOH of aqueous solutions. You will calculate acid and base concentrations and determine concentrations experimentally. You will explain how buffers resist changes in pH.

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Science / Chapter 19 - acids, bases, and salts (handouts)
Chapter 19...Acids, Bases, and Salts ACID-BASE
THEORIES Acids and bases are all around us and part of our everyday life (ex. bodily functions, vinegar, carbonated drinks, citrus fruits, car batteries, antacids, cleaning products, etc.). Acids and bases have the

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following properties: Acids Bases Tart or sour taste
Bitter taste

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