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Chemistry. ... will change the color of an acid-base indicator, and can be strong or weak electrolytes in aqueous solution. Term.

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- Page 689: 3 Answer C,
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a conjugate base is what is left of an ion after it has donated a proton in solution.

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A B; hydrolyzing salts: compounds derived from the reaction of a strong base with a weak acid or a strong acid with a weak base or a solution of a weak base

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Chapter 19 “ Acids,
Bases, and Salts ...

Figure 19.22, page 615.

Neutralization. The
solution of known
concentration is called
the standard solution.
added by using a buret.
Continue adding until
the indicator . changes
color.

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Section 19.1 – Acid-
Base Theories Acids
have a sour taste,
change the color of an
indicator, can be strong

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or weak electrolytes in aqueous solution, and react with metals.

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594 Chapter 19 Acids
and Bases CHAPTER

19 What You ' ll Learn

You will compare acids
and bases and

understand why their
strengths vary. You will

define pH and pOH
and calculate the pH

and pOH of aqueous
solutions. You will

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calculate acid and base concentrations and determine concentrations experimentally. You will explain how buffers resist changes in pH.

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Acids produce what type of ions? H^+ Acids taste ____ Acids turn ____ litmus paper ____.

1) sour 2) blue, red pH

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of an acid is _____ than

7. Less What are two properties of strong acids? Good conductors and strong electrolytes Give an example of at least one strong acid. Any of the following: [...]

Science / Chapter 19 -
acids, bases, and salts
(handouts)

Chapter 19...Acids,

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Bases, and Salts ACID-
BASE THEORIES

Acids and bases are all around us and part of our everyday life (ex. bodily functions, vinegar, carbonated drinks, citrus fruits, car batteries, antacids, cleaning products, etc.).

Acids and bases have the following properties:
Acids Bases Tart or sour taste Bitter taste

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Chapter 19 Acids,
Bases, and Salts -

Science with Mrs ...

According to this Odel,
in an acid-base reaction,
each acid has a (17) ,
and each base has a (18)

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Section Strengths of

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Acid-Base Theories
OBJECTIVES: Define

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