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**Strong Acid +
Strong Base =
Neutral Salt**

**Strong Acid +
Weak Base =
Acidic Salt**

**Weak Acid + Strong
Base = Basic Salt**

**Weak Acid +
Weak Base = any**

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***type depending
on the relative
strengths of the
acid and base
involved.***

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Arrhenius said that acids are hydrogen-containing compounds that ionize to yield hydrogen ions (H^+) in aqueous solution. He also said that bases are compounds that ionize to yield hydroxide

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**ions (OH⁻) in
aqueous
solution.**

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19.1 - Acid-Base**

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Theories. Acids have a sour taste, change the color of an indicator, can be strong or weak electrolytes in aqueous solution, and react with metals. Bases.

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Acids, Bases, and***

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Acids and Bases

PRACTICE

PROBLEMS 1.

Write a balanced formula equation for the reaction that occurs between each of the following pairs of reactants. a. magnesium and

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nitric acid b.

aluminum and

sulfuric acid c.

calcium

carbonate and

hydrobromic acid

d. potassium

hydrogen

carbonate and

hydrochloric acid

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**Acids and bases
come in pairs.**

**General equation
is: $HA(aq) + H_2O(l) \leftrightarrow H_3O^+(aq) + A^-(aq)$**

2. $O(l) \leftrightarrow H_3O^+(aq) + OH^-(aq)$

$O^+(aq) + A^-(aq)$

Acid + Base \leftrightarrow

Conjugate acid +

Conjugate base.

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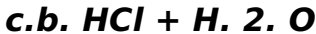
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1-base acid c.a.



1-acid base c.a.

c.b. Amphoteric -

a substance that

can act as both

an acid and

base- as water

shows

Chemistry -

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Acids, Bases, and
Salts - Chapter ...
Acids: taste sour,
react with
certain metals to
produce
hydrogen gas,
react with carbo
nates/bicarbonat
es to produce
carbon dioxide
gas. Corrode
metals, pH is

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**less than 7, turns
blue litmus
paper red.**

**Bases: taste
bitter, have a
slippery
feeling. React
with acids to
form salts and
water, pH is
greater than 7.**

**Science / Chapter
19 - acids, bases,**

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**"Acids, Bases,
and Salts" Tools.**

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account; E-mail
to a friend; Find
other activities
... compounds
derived from the
reaction of a
strong base with
a weak acid or a**

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***strong acid with
a weak base or a
solution of a
weak base and
one of its salts ...
solution that
consists of a
weak acid and
one of its salts:
acid ...***

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Chapter

19...Acids, Bases,

and Salts ACID-

BASE THEORIES

Acids and bases

are all around us

and part of our

everyday life (ex.

bodily functions,

vinegar,

carbonated

drinks, citrus

fruits, car

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**batteries,
antacids,
cleaning
products, etc.).
Acids and bases
have the
following
properties: Acids
Bases Tart or
sour taste Bitter
taste**

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**Thursday May
10, 2012. Acids
taste sour, will
change the color
of an acid-base
indicator, and
can be. strong or
weak
electrolytes in
aqueous
solution. Bases
taste bitter, feel
slippery, will
change the color**

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can be. strong or
weak
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19 “Acids, Bases, and Salts”

Chemistry

**Section 19.1 Acid-
Base Theories**

OBJECTIVES:

**-Define the
properties of
acids and**

**bases. Section
19.1 Acid-Base
Theories**

OBJECTIVES:

-Compare and

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contrast acids and bases as defined by the theories of: a) Arrhenius, b) Brønsted-Lowry, and c) Lewis.

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19 Acids, Bases,
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& Your Blood
Your body
function properly
only when the pH
of your blood lies
between 7.35
and 7.45.

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