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hydrolyzing salts: compounds derived from the reaction of a strong base with a weak acid or a strong acid with a weak ...

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Strengths of Acids and Bases
pages 602-607 BLOCK SCHEDULE

LESSON PLAN 19.2 Objectives

- Relate the strength of an acid or base to its degree of ionization.

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Study Guide: Acids and Bases

1) List at least three characteristic properties of acids and three of bases.

ACIDS pH less than 7 Have a sour taste Change the color of many indicators Are

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corrosive (react with
metals) Neutralize bases
Conduct an electric current
BASES

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what type of ions? H + Acids
taste ___ Acids turn ____
litmus ...

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of acid or base are added. A
buffer system should contain

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considerably more acid than base. Specific buffer systems should be chosen based on the pH that must be maintained. 114 Study Guide for Content Mastery Chemistry: Matter and Change Chapter 19

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Bases PRACTICE PROBLEMS 1.
Write a balanced formula equation for the reaction that occurs between each of the following pairs of reactants. a. magnesium and nitric acid b. aluminum and sulfuric acid c. calcium carbonate and hydrobromic acid d. potassium hydrogen carbonate and hydrochloric acid

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Bossong/Woleslagle/Tang-Johnson Chemistry Study Guide Chapter 19 Study Guide 19.1 Acid-Base Theories

Define the following terms:

- monoprotic, diprotic, triprotic acids - conjugate acids - conjugate bases - conjugate acid-base pairs - hydronium ion (H_3O^+) - amphoteric - Lewis acid - Lewis base 1.

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Quiz & Worksheet - Household Acids, Bases & Salts | Study.com

Chapter 19 Acids, Bases, and Salts Flashcard What are the properties of acids and bases? acids taste sour, will change the color of an acid-base indicator, and can be strong or weak electrolytes in aqueous solution.

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