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2325 Spring 2015 Instructor: Michael  
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AP Chemistry Chapter 22 - Organic  
Chemistry 22.1 Alkanes: Saturated  
Hydrocarbons A. Straight-chain  
Hydrocarbons 1. Straight-chain  
alkanes have the formula  $C_nH_{2n+2}$  2.  
Carbons are  $sp^3$  hybridized The First  
10 Alkanes # of Carbons Name  
Formula ( $C_nH_{2n+2}$ ) 1 Methane  $CH_4$  2  
Ethane  $C_2H_6$  3 Propane  $C_3H_8$  4  
Butane  $C_4H_{10}$  5 Pentane  $C_5H_{12}$  6  
Hexane  $C_6H_{14}$  7 Heptane  $C_7H_{16}$

## Chapter 22

Chapter 22 Organic Chemistry.  
Organic Chemistry. The study of  
compounds that contain the element  
carbon. Organic molecules contain  
carbon. Hydrocarbons ( $CH_4$ ,  $C_3H_8$ ,  
 $C_6H_6$ ), various acids ( $C_2H_4O_2$ )  
2)Pharmaceuticals, clothes, gasoline,

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plastic, nearly everything inside of a living organism, etc.

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22. Nucleophilic Addition-Elimination  
Reactions ... this text, we will  
sometimes refer to this component as  
the 'acyl X' group (this, however, is  
not a standard term in organic  
chemistry). Notice that the acyl X  
groups are simply deprotonated forms  
of other functional groups: in an  
amide, for example, the acyl X group



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is an amino group ...

## **Chapter 22 Notes**

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Chapter 22: Carbonyl Alpha-

Substitution Reactions Last updated;

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headers. Alpha?substitution reactions

are the third major type of reaction that

you will study in your investigation of

the chemistry of carbonyl compounds.

... The chapter concludes ...

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### **Flashcards and Study Sets ...**

Chapter 22: Organic Chemistry 4.

Refer to Sections 22.1 and 22.2 and Example 22.2. For a hydrocarbon with  $n$  carbons, the number of hydrogens in an alkane equals  $2n + 2$ , in alkene it equals  $2n$ , in an alkyne it equals  $2n - 2$ . This assumes the hydrocarbons are not cyclic and contain at most, one double bond. a.

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