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who discovered radioactivity? Becquerel, Marie Curie, Pierre Curie  
radioisotope form of an atom that is unstable & loses particles from the nucleus  
radioactivity penetrating rays & particles given off by unstable nuclei  
half-life the amount of time for 1/2 the radioactive nucleus to decay  
nuclear FISSION large nuclei is split into two smaller nuclei  
nuclear [...]

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Nuclear Energy Section 4.1 Early Ideas About Matter For each statement below, write true or false. ... 37X 36.966 24.23 Section 4.4 Unstable Nuclei and Radioactive Decay For each item in Column A, write the letter of the matching item in Column B.

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1.04 - NUCLEAR PHYSICS RCT STUDY GUIDE-3-  
Issued 05/95 1.04.02 Identify the basic principles of the mass-energy equivalence concept. MASS-ENERGY EQUIVALENCE Of fundamental concern in nuclear reactions is

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the question of whether a given reaction is possible and, if so, how much energy is required to initiate the reaction or is released when

### **CHAPTER 24 - NUCLEAR CHEMISTRY**

CHAPTER 24 NUCLEAR REACTIONS AND THEIR APPLICATIONS 24.1 a) Chemical reactions are accompanied by relatively small changes in energy while nuclear reactions are accompanied by relatively large changes in energy. b) Increasing temperature increases the rate of a chemical reaction but has no effect on a nuclear reaction. c) Both

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chemical and ...

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24 CHAPTER Section 24.3 continued Heat produced by nuclear fission is carried away

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by (24) Date Class STUDY GUIDE , which enters the core at point (25) point (26) in the diagram. It then leaves the core at Heat from the reactor core is used to boil water in the (27) shown at (28) generate electricity at point (29) cooled at location (30)

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284 Study Guide for An Introduction to Chemistry Section Goals and Introductions Section 18.1 The Nucleus and Radioactivity Goals To introduce the new terms nucleon, nucleon number, and nuclide. To show the symbolism used to represent nuclides. To

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