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s and Changes
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71) Mixture Two

or more

substances,

combined in

varying

proportions -

each retaining

its own specific

properties. The

components of a

mixture can be

separated by

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making and

breaking of

chemical bonds.

(Phys.

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and ...

Name _____

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3 !! III.
Mixtures of Iron

Matter

(Classifying
Matter) pgs.

66-69 a.

Substance - made
of only _____
kind of atom or
molecule b.

Mixtures -
combination of
_____ pure
substances in

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which each pure substance retains its individual chemical properties.

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Reviewing

Vocabulary 1. f

2. d 3. e 4. a

5. c 6. b 7. i

8. g 9. h 10. k

11. j 12. Both

are

characteristics

of substances. A

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Chmielewski 29

Terms.

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properties and
chemical
properties of
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to complete the
passage.

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Changes
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DRAFT. 11th

grade. 62 times.

Chemistry. 83%

average

accuracy. a

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kbonifas. 0.

Save. Edit. ...

A form of matter

with a constant

volume that

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It takes the shape
of its
container.

answer choices .

law of

conservation of

mass. solid.

liquid.

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chemical
properties of
matter. Use each
of the terms
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(1) and volume.

A (2) is a form of matter with a uniform and unchanging composition.

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and Changes Quiz

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Energy Chapter

Summary Chapter

3 addresses

links between

the macroscale

properties of

matter and the

properties of

its particles.

Different states

of matter and

types of energy

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are defined.

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...

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Properties And
Changes 1.

Chapter 3
Matter and
Changes
 2.

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Properties of
Matter (3.1)
Matter is

anything that

takes up space

and has mass.
Mass is the

amount of matter

in an object.

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—Properties and

Changes

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Figure 3-1

Miners relied on
the physical
property of
density to
distinguish
gold (19 g/cm^3)
from the
worthless

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minerals in
their sluice
pans. The
density of
pyrite, a
worthless
mineral often
mis-taken for
gold, is 5
g/cm³. Physical
Properties of
Common
Substances State
Melting Boiling

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Density

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Elements and
Compounds g

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12. 13.

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substance change
another
substance.

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physical and
chemical
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This video summarizes the classification of matter, physical properties, chemical changes, conservation of mass, law of definite proportions and separation of a mixture.

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CHAPTER 3 Visit
the Chemistry
Web site at science.glencoe.com
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about matter.

Chemistry is the
study of matter

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and its
properties.

Every aspect of these divers' environment, under water and on land, is some form of matter.
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