

Chapter 6 Chemistry Review

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f. mass of 6 mol C = 6 (12.01 g) = 72.06 g mass of 6 mol H = 6 (1.008 g) = 6.048 g mass of 1 mol O = 16.00 g = 16.00 g molar mass of C₆H₆O = 94.11 g 21. a. molar mass of SO₃ = 80.07 g 49.2 mg SO₃ 1 g 1000 mg 1 mol 80.07 g = 6.14 10⁻⁴ mol SO₃ b. molar mass of PbO₂ = 239.2 g 7.44 x 10⁴ kg PbO₂ 1000 g 1 kg 1 mol 239.2 g = 3.11 10⁵ mol PbO₂

General Chemistry 1 Review Study Guide - IB, AP, & College Chem Final Exam

View Homework Help - Chapter 6 Review Answers from CHEMISTRY CHEM1B at Laney College. CHAPTER PERIODIC TABLE 6.1 Concept Review page 149 1. For the lanthanoid and actinoid elements, proposed a

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Chapter 6 Review

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CHAPTER 6 REVIEW Chemical Bonding SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.

Chapter 6 Chemical Composition - Francis Howell High School

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6 Chemical Bonding

AP CHEMISTRY CHAPTER REVIEW CHAPTER 6: ELECTRONIC STRUCTURE AND THE PERIODIC TABLE You should be familiar with the wavelike properties of light: frequency (), wavelength (), and energy (E) as well as the equations that show their relationships (E = h and c =) You should be familiar with the electromagnetic spectrum and atomic spectra (bright-line spectra).

CHAPTER 6 ANSWERS AND EXPLANATIONS - Chemical Reactions ...

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