

Chapter 7 Review Modern Chemistry Answers

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered.

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7 Chemical Formulas and Chemical Compounds

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO₃ a. copper(II) carbonate Na₂SO₃ b. sodium sulfite (NH₄)₃PO₄ c. ammonium phosphate SnS₂ d. tin(IV) sulfide HNO₂ e. nitrous acid 2.

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on ...

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HOMEWORK 7-2

CHAPTER 6 REVIEW Chemical Bonding MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. a.

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Holt McDougal Modern Chemistry Chapter 7: Chemical ...

Modern Chemistry • CHAPTER 7 HOMEWORK 7-2 ... CHAPTER 7 • HOMEWORK 7-2 VOCABULARY 1. d 2. c 3. e 4. a 5. g 6. h 7. b 8. f GRAPHIC ORGANIZER Name of Compound or Ion Formula of Compound Copper(II) ion Cu^{2+} Ammonium ion NH_4^+ Bismuth(III) bromide BiBr_3 Stannous fluoride SnF_2 Magnesium fluoride MgF_2 Iron(III) cyanide $\text{Fe}(\text{CN})_3$

5 The Periodic Law

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds

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7 Chemical Formulas and Chemical Compounds

Modern Chemistry 57 Chemical Bonding CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Label each of the following statements as True or False: _____ a. If the formula mass of one molecule is x u, the molar mass is x g/mol. _____ b.

6 Chemical Bonding

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