

Chapter 8 Covalent Bonding Workbook Answers

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Section 8.4 – Polar Bonds and Molecules. Covalent bonds involve sharing electrons between atoms. When the atoms in the bond pull equally, the bonding electrons are shared equally, and the bond is nonpolar. When the atoms in the bond pull unequally, the bonding electrons are pulled closer to one atom, and the bond is polar.

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Chapter 8 Covalent Bonding and Molecular Structure 8-4 H 2 molecule. More sophisticated descriptions of chemical bonding will be discussed in Chapter 9. 8.3 Lewis Structures OWL

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Opening Exploration 8.X One of the most important tools chemists use to predict the properties of a chemical species is its Lewis structure.

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1. In covalent bonds, electron sharing usually occurs so that atoms attain the electron configuration of noble gases. (8) 2. Atoms form double or triple covalent bonds if they can attain a noble gas structure by sharing two pairs or three pairs of electrons.

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242 Chapter 8 • Covalent Bonding Single Covalent Bonds When only one pair of electrons is shared, such as in a hydrogen molecule, it is a single covalent bond. The shared electron pair is often referred to as the bonding pair. For a hydrogen molecule, shown in Figure 8.4, each covalently bonded atom equally

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COVALENT BONDING Class Name Date COVALENT BONDING Class 8.2 8.2 8.4 8.3 8.3 8.3
195 Vocabulary Review Select the term from the following list that best matches each description. e,chapter Quiz loose the best answer and write its letter on the line. . A bond in which each atom contributes two electrons is ... Chapter 8 Covalent Bonding .

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The bonding pairs of electrons in covalent bonds are pulled, as in the tug-of-war in Figure 8.2 between the nuclei of the atoms sharing the elec-trons. When the atoms in the bond pull equally (as occurs when identical atoms are bonded), the bonding electrons are shared equally, and the bond is a nonpolar covalent bond. Molecules of hydrogen (H

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Match. Gravity. Created by. Petitt1. Terms in this set (30) covalent bond. a chemical bond that involves sharing a pair of electrons between atoms in a molecule. molecule. two or more atoms held together by covalent bonds.

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