

Chapter Test B Stoichiometry

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Example 1. How many molecules of SO 3 are needed to react with 144 molecules of Fe 2 O 3 given this balanced chemical equation? Fe 2 O 3 (s) + 3SO 3 (g) ? Fe 2 (SO 4) 3. Solution. We use the balanced chemical equation to construct a conversion factor between Fe 2 O 3 and SO 3.The number of molecules of Fe 2 O 3 goes on the bottom of our conversion factor so it cancels with our given amount ...

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Chapter Test B Chapter: Stoichiometry PART I Inthe space provided, write the letter ofthe term or phrase that best completes each statement or bestanswers each question. 1. Knowingthe mole ratio ofa reactantandproductin a chemical reaction would allowyoutodetermine a. the energyreleased inthe reaction. b. thespeedofthe reaction.

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Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

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