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Keywords: A-level Chemistry; 2420; ER; CHEM4; ; 2013; Examiner report Created Date: 10/10/2013 9:09:58 AM

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Mark Scheme – General Certificate of Education (A-level) Chemistry – Unit 5: Energetics, Redox and Inorganic Chemistry – January 2011

Question Marking Guidance Mark Comments electrons 3(a) Na 2 O ionic Strong forces between ions/strong ionic bonding SiO 2 macromolecular
Strong covalent bonds (between atoms) 1 1 1 1

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Chemistry - AQA GCE Mark Scheme 2010 January series 8 Question Part Sub Part Marking Guidance Mark Comments 5 (a) A solution v
resists change in pH when small amounts of acid or base are added / on dilution NH₄Cl (allow other correct strong acid) 1 1 1 Do not
remains constant 5 (b) (i) $K_a = [H$

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Mark Scheme - General Certificate of Education (A-level) Chemistry - Unit 4: Kinetics, Equilibria and Organic Chemistry - January 2013 9 Question Marking Guidance Mark Comments 5(a) (i) Single/one (intense) peak/signal OR all H or all C in same Do n environment OR 12 or 4 equiv C OR

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5 A2 AQA Chemistry 2420 January 2013 CHEM2 Q09. A2 AQA Chemistry 2420 January 2013 CHEM2 Q09. Mark Scheme - General Certificate of Education (A-level) Chemistry - Unit 2: Chemistry In Action - January 2013 23 . Question . Marking Guidance . Mark . Comments . 9 . M2 (either order) Any two from

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WMP/Jan13/CHEM2 Do not write outside the box 4 (04) 13 1(c) Barium metal reacts very quickly with dilute hydrochloric acid, but it reacts slowly with water. 1 (c) (i) Write an equation for the reaction of barium with water. (1 mark) 1 (c) (ii) A solution containing barium ions is used to show the presence of sulfate ions in an aqueous solution of sodium sulfate.

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