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Six Types of Reaction Practice Sheet Answers. Using your knowledge of the six types of chemical reaction, balance the following equations and indicate which of the six types of reaction

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they are: 1) $2 \text{LiH} + 1 \text{F}_2 \rightarrow 2 \text{LiF} + 1 \text{H}_2$. Type of reaction: Single replacement. 2) $2 \text{Fe}(\text{OH})_2 \rightarrow 1 \text{Fe}_2\text{O}_3 + 3 \text{H}_2\text{O}$ Type of reaction: Decomposition.

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Chemistry

Reaction type: _____ combustion 5)
 $2 \text{PbSO}_4 \rightarrow 2 \text{PbSO}_3 + 1 \text{O}_2$ Reaction
type: _____ decomposition 6) 2NH_3
 $+ 3 \text{I}_2 \rightarrow 1 \text{N}_2 + 6 \text{HI}$ Reaction

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type:_____single displacement 7) 1

$H_2O + 1 SO_3 \rightarrow 1 H_2SO_4$ Reaction

type:_____synthesis 8) $H_2SO_4 + 2$

$NH_4OH \rightarrow 2 H_2$

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M. Types of reaction (4) N.

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Predicting rxn products (5) O.
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Balancing Equations Practice
Worksheet

Types of Reactions Worksheet –
Solutions Balance the following
equations and indicate the type of
reaction taking place: 1) $3 \text{NaBr} + 1$
 $\text{H}_3\text{PO}_4 \rightarrow 1 \text{Na}_3\text{PO}_4 + 3 \text{HBr}$ Type of

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reaction: double displacement 2) $3 \text{Ca(OH)}_2 + 1 \text{Al}_2(\text{SO}_4)_3 \rightarrow 3 \text{CaSO}_4 + 2 \text{Al(OH)}_3$ Type of reaction: double displacement 3)

Quiz #2-1 PRACTICE: Types of Chemical Reactions | Mr ...

This Types of Reaction Quiz tests

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you on the key terms of various types of chemical reactions such as combination, decomposition, displacement, etc. Identify chemical reactions as either synthesis, combustion, decomposition, single or double replacement.

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- ProProfs Quiz

Six Types of Chemical Reaction
Worksheet Answers. Balance the
following reactions and indicate
which of the six types of chemical
reaction are being represented: a) 2.
NaBr + 1. Ca(OH)₂ (1. CaBr₂ + 2.

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NaOH. Type of reaction: double displacement. b) 2. $\text{NH}_3 + 1. \text{H}_2\text{SO}_4 \rightarrow 1 (\text{NH}_4)_2\text{SO}_4$. Type of reaction: synthesis. c) 4. $\text{C}_5\text{H}_9\text{O} + 27. \text{O}_2 \rightarrow 20. \text{CO}_2 \dots$

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Types of Reactions Worksheet

Indicate the type of reaction taking place:

1) $\text{NaBr} + \text{H}_3\text{PO}_4 \rightarrow \text{Na}_3\text{PO}_4 + \text{HBr}$ Reaction type: _____

2) $\text{Ca(OH)}_2 + \text{Al}_2(\text{SO}_4)_3 \rightarrow \text{CaSO}_4 + \text{Al(OH)}_3$ Reaction type: _____

3) $\text{Mg} + \text{Fe}_2\text{O}_3 \rightarrow \text{Fe} + \text{MgO}$ Reaction type: _____

4) $\text{C}_2\text{H}_4 + \text{O}_2$

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Types of Reactions Worksheet
THEN Balancing!. First, begin by telling which type of reaction is taking place. Then go back and balance the following . equations:

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The six types of reaction. You may wonder why this is something that's important, and frankly, that's not a bad question. You see, the six

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types of reaction are only one of many different ways to categorize chemical reactions. For example, your teacher may teach you things like "redox reaction" and "nuclear reaction" as types of reaction,...

Types of Chemical Reaction

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WORKSHEET ANSWERS
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Effect of Prior Experience, Age, and Gender in Non-Food Blending Predictions. Though this title sounds pretty scientific, it just refers to an experiment I did with putting rubber balls in a blender to see...

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