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**2.2 MASS-ACTION KINETICS**

**2.2.1 OVERVIEW We assume  
that rates can be described  
by  $\text{Rate} = (k) \cdot (C^m)$  where  $(k)$  is  
the rate constant from  
Arrhenius' law and  $(C^m)$  is the  
rate law. Empirically, we say  
that  $(C^m) = \dots$  which is typical  
rate law kinetics.**

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