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Find the equilibrium concentration of one of the reactants in a reaction when the equilibrium expression is a perfect square.  
Find the equilibrium concentration when the equilibrium constant is given and the equilibrium expression is **NOT** a perfect

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square; you will use the quadratic equation.

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Students should be able to... Describe a .  
system in equilibrium . Find the reaction .

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quotient,  $Q$ , ( $K_{eq}$ ) for a reaction equation .  
Solve for  $Q$  given . initial concentrations or  
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There are three main types of equilibrium: chemical equilibrium, phase equilibrium, and solution equilibrium. Chemical Equilibrium As in the previous examples, the products of a reaction can go in reverse,

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creating an equaling out of products and reactants.

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Guide When you see a chemical reaction  
formula what represents: A reaction in  
equilibrium A forward only reaction What

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three things can adjust the equilibrium to shift either right or left? What does NOT change while a system is in equilibrium? or What does "Nature of Reactants" refer to?

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Use this lesson plan to teach your students about chemical equilibrium. Students will

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watch a video lesson that defines the term, explains how it's dynamic, tells about equilibrium constant, and ...

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This chemistry video tutorial provides a basic introduction into Le Chatelier's Principle of chemical equilibrium. It



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explains how to determine which direction the reaction will shift if the ...

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equilibrium is when the rate of a forward

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reaction equals the rate of the reverse reaction and the concentrations of the products and reactants remain unchanged....

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properties of substances. Now methods or novel analytical techniques are emerging for rapid and reliable measurements of these physical and chemical properties.

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Chemical Equilibrium A state in which the



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rates of forward and reverse reactions are the same. An active, dynamic condition, not static. All substances are being made & unmade at the same rate, so their concentrations are constant.

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