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equation A representation of a chemical reaction with the formula of th... Skeleton equation A chemical equation that does not indicate the relative amount... A representation of a chemical reaction that uses symbols to s... Shows the formulas for reactants...

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Date Chapter Test B, continued 17. A

substance combines with oxygen, releasing a
large amount of energy as heat and light, in

a(n) . 18. The decomposition of a substance

by an electric current is called . 19. A(n)

orders the elements by the ease with which
they undergo certain chemical reactions. 20.

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a substance that increases the rate of a chemical reaction without itself undergoing any permanent chemical change. catalyst
A equation where each side of the equation has the same number of atoms of each element, and mass is conserved

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Chemical Reactions Test When writing chemical equations, simply type subscripts in as numbers. Do not leave spaces between formulas and symbols and type > (usually "Shift .") for an arrow. For example to write the

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balanced chemical equation for hydrogen reacting with oxygen to produce water type $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$.

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The chemical reactions that occur inside the human body are extremely complex. One of the main reactions that occurs is the metabolism of glucose, and another reaction is the

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detoxification of the liver. The metabolism of glucose is one of the main factors that affect the levels of insulin that are generated by the liver.

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Skeleton equation O. chemical equation I. a new substance formed in a chemical reaction a starting substance in a chemical reaction an equation in which each side has the same number of atoms of each element tale/11>

Multiple Choice Identify the letter of the choice that best completes the statement or

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answers the question.

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Chemical equation A representation of a chemical reaction with the formula of th...
Skeleton equation A chemical equation...

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chem 11 and 12. A ----- is a chemical equation that does not indicate the amount of reactants and products.

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a reaction in which a single compound is broken down into simpler substances
single-replacement reaction a reaction in which the atoms of one element replace the atoms of a second element

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Chapter 11" ...

Chapter 11 Practice Test Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question. _____ 1. Chemical reactions _____.

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