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ones of the following

reactions are endothermic in

other words ΔH is positive?

I. $\text{H}_2\text{O}(\text{l}) + 10,5\text{kcal} ?$

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expand the treatment
gradually into progressively
...

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Example 9.2 The element mercury, Hg, is a silvery liquid at room temperature.

The normal freezing point of mercury is -38.9°C , and its

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molar enthalpy of fusion is
 $\Delta H_{\text{fusion}} = 2.29 \text{ kJ/mol}$. What
is the entropy change of the
system when 50.0 g of Hg(l)
freezes at the normal
freezing

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application of chemical thermodynamics to the metals & materials which later on known as Thermodynamics of materials. 1.1.1 Importance of Thermodynamics: 1.

Stoichiometry and

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