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Chapter 13 states of matter pearson chemistry. in a system at constant vapor pressure, a dynamic equilibrium exists between the vapor and the liquid. The system is in equilibrium because the rate of evaporation of liquid equals the rate of condensation of vapor.

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Chemistry - Chp 13 - States of Matter. 16. Section 13.1 The Nature of Gases? Equal pressures:1 atm = 760 mm Hg = 101.3 kPa? Sample 13.1, page 387? Most modern barometers do not contain mercury- too dangerous ? These are called aneroid barometers, and contain a sensitive metal diaphragm...

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Chemistry Chapter 13: States Of Matter Review. Match the intermolecular forces with their descriptions. 1. Weak forces between nonpolar molecules. 2. A type of one of the forces that is between hydrogen and a negatively charged particle. 3. Attractions between oppositely charged regions of polar molecules.

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