

Chemistry Chapter 9 Stoichiometry Test

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equation. Chapter 9 Stoichiometry Table of Contents Chapter 9 Stoichiometry Section 1 Chapter 9 Section 1 Intro to Stoichiometry. STOICHIOMETRY Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Blair12_Armstrong. Key Concepts: Terms in this set (10) Stoichiometry is a branch of chemistry that deals with elements in compounds ...

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Practice Test Ch3 Stoichiometry (page 3 of 3) 1. d It might be easiest to balance the equation with mostly whole numbers: 2 NH₃ + ? O₂ → 3H₂O. The question asks about the amount of oxygen reacting with ONE mole of ammonia, thus cut the ??? (3.5) of oxygen in half to 1.75. Balance: CH

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CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation: C₃H₄(g) + xO₂(g) → 3CO₂(g) + 2H₂O(g) 4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar mass of C₃H₄? 2 mol O₂:1 mol H₂O c. What is the mole ratio ...

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To understand stoichiometry, start with this introduction to the topic. It might also help to review molecules and moles, which includes the formulas work. Ready for another quiz? Here's a quick self-test about the mole. If you'd rather switch gears, see if you know the answers to these questions. chemistry explains the real world.

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