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9

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Match each item with the correct
statement below. Match each item with
the correct statement below. a.

monatomic ion f. cation b. acid g. binary
compound c. base h. anion d. law of

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definite proportions i. polyatomic ion e.
law of multiple proportions _____ 1.

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Chapter 9: Stoichiometry ...
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explore and make use of the seemingly
limitless changes that matter can
undergo, chemists and chemistry students
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CHAPTER 9 REVIEW Stoichiometry

SECTION 3 PROBLEMS Write the

answer on the line to the left. Show all
your work in the space provided. 1. 88%

The actual yield of a reaction is 22 g and
the theoretical yield is 25 g. Calculate the
percentage yield. 2. 6.0 mol of N_2 are
mixed with 12.0 mol of H

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