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Chemistry IF8766 9 OInstructional Fair, Inc. 59. CALCULATIONS USING SIGNIFICANT FIGURES ... The answer is expressed as 190,000 cm³ since 19 cm has only two significant fiaures. I I When adding and subtracting, limit and round your answer to the least number of decimal places in any of the numbers that make up your answer.

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LE CHATELIER'S PRINCIPLE Name - Ms. Ose's Chemistry Website

MOLES AND MASS Name Determine the number of moles in each of the quantities below. l. 25 g of Naa 2. 125 g of 3. 100. g of 4 74 g of KCl z LB

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Name: _____ Naming Ionic Compounds Directions: Name the following compounds using the Stock Naming System. 1. CaCO 3 calcium carbonate_____ 2. KCl potassium chloride ...

SIGNIFICANT FIGURES Name

Writing Formulas From Names Directions: Write the formulas of the following compounds. 1. ammonium phosphate (NH 4) 3 PO 4 _____ 2. iron (II) oxide FeO_____ 3. iron (III) oxide Fe 2 O 3 ... Chemistry IF8766 Page 48 Instructional Fair, Inc. Title: Name:

Writing Formulas From Names - Ms. Agostine's Chemistry Page

IDEAL GAS LAW Use the ideal Gas Law below to solve the following problems. pressure in atmospheres volume in liters number of moles L atm Universal Gas Constant = 0.0821

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KMBT 654-20130920111026 - Berger's Chemistry Class

Le Chatelier's Principle states that when a system at equilibrium is subjected to a stress, the system will shift its equilibrium point in order to relieve the stress. Complete the following chart by writing left, right or none for equilibrium shift, and decreases, increases or remains the same for the concentrations of reactants and

Combined Gas Law Worksheet Answer Key Instructional Fair

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Redox Reactions Worksheet Instructional Fair

r i c Page 4:2 ANSWER KEY WRITING FORMULAS (CRISS-CROSS METHOD) Name IONIC BONDING Name _ Ionicbondingoccurswhenametcl hcnstersonemoreelectrons10anonmeloln on ...

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Answers: COMBINED GAS LAW Remember to convert all temperatures to Kelvin. P 1 V 1 T 1 P 2 V 2 T 2 1 1.5 atm 3.0 L 20. C 293K 2.5 atm 1.9 L 30. C 303K 2 720 torr 256 mL 25 C 298 K 8.0x102 torr 250 mL 50. C 323 K 3 600. mmHg 2.5 L 22 C 295 K 760 mmHg 1.8 L 270 K 4 1.2 atm 750 mL 0.0 C 273.0 K 2.0 atm 500. mL 25 C 298 K 5 95 kPa 4.0 L

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