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Study Guide 20 Chemistry: Matter and Change Chapter 9 . Name CHAPTER Section 9.1 continued Date Class STUDY GUIDE For each of the following chemical reactions, write a word equation, a skeleton equation, and a balanced chemical equation. Be sure to show the state of each reactant and prod-

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As you begin your study of chemistry—the study of matter and the changes that it undergoes—you are probably asking yourself, “Why is chemistry important to me?” The answer to this question can be illustrated by real-life events that involve two discoveries. One discovery involves something that you probably use every day—refrigeration.

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the study of matter and the changes it undergoes matter that has a definite and uniform composition a substance that consists of chlorine, fluorine, and carbon a measurement that reflects the amount of matter

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octet rule atoms tend to gain, lose, or share electrons in order to acquire eight valence electrons chemical bond the force that holds two atoms together cation positively charged ion anion negatively charged ion compound a chemical combination of two or more different elements ionic bond the electrostatic force that holds oppositely charged particles together [...]

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Glencoe Chemistry - Matter And Change Chapter 13: Gases ...

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